

Combined Directed Ortho Metalation/Cross-Coupling Strategies: Synthesis of the Tetracyclic A/B/C/D Ring Core of the Antitumor **Agent Camptothecin**

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A convergent synthesis of the A/B/C/D ring fragment 5 of camptothecin using a combination of directed ortho metalation and Negishi cross-coupling is described. The key features of the synthetic sequence are an anionic ortho-Fries rearrangement (10 \rightarrow 12), a Negishi cross-coupling (7 \rightarrow 6), and a terminal modified von Braun reaction $(16 \rightarrow 5)$ that leads to tetracyclic derivative 5 in 7 steps and 11% overall yield.

Introduction

(20S)-Camptothecin (1a), one of the most potent antitumor natural products, was first isolated in 1966 from Camptotheca acuminata Nyssaceae by Wall et al.^{1,2} Its unusual structure and intriguing speculative biosynthetic pathway³ stimulated immediate and intense synthetic activity worldwide, 4,5 which subsequently precipitously declined due to the finding of discouraging biological activity data of 1a⁵ showing serious toxic side effects. In the past 15 years, renewed synthetic interest in 1a and analogues⁶ evolved due to findings that substituted derivatives of **1a** such as 9-aminocamptothecin (**1b**). 9-((dimethylamino)methyl)-10-hydroxycamptothecin (Topotecan, 1c),8 and Irinotecan (1d)9 show low overall toxicity, higher solubility, and still impressive in vivo activity against certain solid tumors. In fact, Topotecan 1c and Irinotecan 1d were recently approved by the FDA

for the treatment of ovarian cancer and small-cell lung cancer¹⁰ and refractory colorectal cancer, respectively.¹¹

1a:
$$R_1$$
, R_2 , R_3 , R_4 = H, (20S)-Camptothecin
1b: R_1 = NH₂, R_3 , R_4 = H
1c: R_1 = CH₂NMe₂· HCl, R_2 = OH, R_3 = H
Topotecan
1d: R_1 , R_3 = H, R_2 = OCON
 R_4 = Et
Irinotecan

Alkaloids structurally related to camptothecin (1a) such as homocamptothecin (2), 12g,13 mappicine (3), 14,15 and mappicine ketone $(4)^{15}$ are also of clinical relevance. Homocamptothecin (2) and its derivatives show similar therapeutic activities to the camptothecins (1). The E-ring lactone in 1a easily hydrolyzes at physiological pH leading to the biologically inactive carboxylate, an inherent deficiency of the parent compound 1a.16 In the E-ring lactone of 2 the tertiary alcohol and the lactone carbonyl are separated by a methylene spacer. It was found in bioassays that the longevity of activity of 2 was increased compared to that of 1a.17 Mappicine ketone (4) has been identified as an antiviral lead compound with selective

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FIGURE 1. Selected strategies for the synthesis of camptothecin.

activities against herpes viruses HSV-1 and HSV-2 and human cytomegalovirus. 18

2: Homocamptothecin

3: R=OH, R'=H (Mappicine) 4: R, R'=O (Mappicine ketone)

The antitumor activity of the camptothecins is now accepted to be associated with the inhibition of DNA relaxation by specific interference of the function of DNA topoisomerase I.^{19,20} Interestingly, it has been shown that the tetracyclic A/B/C/D ring core of **1a** functions as the key binding site to DNA.²¹

Camptothecin and its analogues have provided a rich playing field for development of convergent total synthesis strategies. To date, the shortest asymmetric synthesis of **1a** by Comins involves the formation of the C-ring by connecting the A/B- and D/E-fragments via an N-alkylation and a key intramolecular Heck ring closure reaction (Figure 1, A).22 Curran devised an imaginative strategy in which the appropriately functionalized A- and D/E-fragments (Figure 1, B) participate in a free-radical cascade leading to the formation of the B- and C-rings of 1a.12 A different concomitant formation of the B- and C-rings was reported by Fortunak using an efficient intramolecular Diels-Alder reaction (Figure 1, C) that is now used on an industrial scale.23 Boger devised an approach in which a D-ring forming intermolecular Diels-Alder process precedes a C-ring cyclization (Figure 1, D).¹⁷ The key features of an inventive strategy²⁴ by Bosch consisted of an intramolecular radical cyclization to form the C-ring followed by asymmetric construction of the E-ring using enolate chemistry (Figure 1, E).

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SCHEME 1

SCHEME 2

1. NaH / THF
$$0^{\circ}C \rightarrow rt$$

2. CICONEt₂ reflux / 3h (88%)

10

1. LDA / THF -75°C / 1h 2. MeOH

12 (61%)

13 (13%)

Tf₂O / Net₃ CH₂Cl₂ / 0°C (70%)

CONEt₂

CONEt₂

CONEt₂

CONEt₂

Tf₂O / Net₃ CH₂Cl₂ / 0°C (70%)

Intermolecular²⁵ (Figure 1, F) and intramolecular²⁶ (Figure 1, G) Michael addition reactions by Ciufolini and Chavan, respectively, were utilized for the D-ring construction of **1a**. Finally, in a route that has been widely traveled from the beginning of camptothecin synthetic work, a classical Friedlander reaction of 2-aminobenzaldehyde (A-ring) with the assembled C/D/E-ring framework (Figure 1, H) was used by Henegar for B-ring annelation of 1a.11

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Herein, we report a new synthesis of the A/B/C/D-ring core 5^{27} of 1a using a combined directed ortho metalation (DoM)—transition metal catalyzed cross-coupling tactic.²⁸ Thus, the construction of 5 is based on the initial coupling of **7**, prepared by anionic Fries rearrangement, ²⁹ with the organozinc species 8 to afford 6, which, by simple functional group manipulation-cyclization leads to Cring formation (Scheme 1).

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Results and Discussion

As an appropriate precursor of coupling partner 7, 2-quinolone (9) (Scheme 2), prepared in sizable quantities following Henze's procedure, 30 was converted into the O-carbamate 10 by treatment with sodium hydride and diethyl carbamoyl chloride in refluxing THF³¹ followed by purification by flash chromatography on SiO₂. In an alternative purification of carbamate 10 via short path distillation, a clean thermal 1,3-carbamoyl rearrangement was observed to form urea 11.32 As reported by Queguiner, 31 treatment of carbamate 10 with LDA resulted in, even at -75 °C, C-3 metalation—anionic Fries rearrangement to give the 3-amidoquinolone 12. However, in contrast to the observations of Queguiner, apart from **12** (61%), the self-condensation product **13** was also isolated in 13% yield. When the reaction was carried out at -42 °C, compound 13 was not formed and the yield of the desired rearrangement product **12** increased to 71%. The increased selectivity favoring the intramolecular rearrangement at higher temperature may be rationalized by the difference in temperature dependence of a

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SCHEME 3

uni- vs. bimolecular reaction. To complete the synthesis of the A/B-fragment, the quinolone 12 was transformed into the triflate 7 with use of standard conditions.

As a model reaction, the cross-coupling reaction between triflate **7** and 2-bromopyridine, 33 representing the D-ring fragment, was undertaken (Scheme 3) with use of the Negishi protocol 34 with Pd(PPh₃)₄ as catalyst and provided the biaryl **14** in satisfactory yield (57%). With this procedure, 2-bromo-6-methoxypyridine, prepared from 2,6-dibromopyridine (86% yield), 35 was sequentially treated with 2 equiv of t-BuLi at -78 °C and anhydrous ZnBr₂. The resulting organozinc species **8** was subjected to the Pd⁰-catalyzed cross-coupling procedure with triflate **7** to afford the biaryl **6** in 59% yield.

To avoid complications of reduction of π -deficient heterocyclic rings, the biaryl **6** was subjected to the mild reduction protocol of Raucher.³⁶ Thus, **6** was converted with use of Lawesson's reagent into the corresponding thioamide **15**, which, upon sequential ethylation with ethyl-Meerwein salt and reduction with NaBH₄, was transformed into the tertiary amine **16** in 83% yield. The final cyclization of **16** to the tetracycle **5**²⁷ was achieved in 62% yield via a modified von Braun reaction with ethyl chloroformate and potassium carbonate.³⁷ Compound **5** was found to be identical by comparison of physical and spectroscopic properties with those reported (see the Experimental Section).

Thus, the tetracyclic A/B/C/D-ring system **5** of camptothecin (**1a**) has been synthesized from quinolone **9** in 7 steps and 11% overall yield, which represents a valuable alternative to previously achieved syntheses (7 steps, 6%;^{27a} 2 steps, 27%;^{27b} and 5 steps, 17%^{27c}). The simplicity of the steps, the ready availability of starting materials of both A/B- and D-ring fragments by DoM

chemistry, and the potential further modification of ring D in 5, taken in sum, offer consideration of additional avenues for synthetic excursions in the camptothecin field.

Experimental Section

N.N-Diethyl O-(Quinolyl-2) Carbamate (10). To a suspension of NaH (959 mg, 60% in mineral oil, 24.0 mmol) in anhydrous THF (45 mL) was added 9³⁰ (2.88 g, 19.8 mmol) portionwise at room temperature under N₂ atmosphere. The greenish suspension was stirred at room temperature for 30 min and then ClCONEt2 (3.75 g, 27.6 mmol) was added via syringe. The reaction mixture was refluxed for 2 h. Since TLC showed only about 50% conversion, more ClCONEt₂ (3.75 g, 27.6 mmol) was added and the mixture was stirred for an additional hour. The reaction mixture was cooled in ice and carefully treated with saturated aq NH_4Cl (20 mL). The aqueous layer was separated and extracted with Et₂O (2 \times 20 mL). The combined organic phase was washed with H₂O (50 mL) and brine (50 mL) and dried (Na₂SO₄). Removal of the solvent gave the crude product as a brown oil, which was purified by flash chromatography on SiO₂ (Et₂O/hexane (1:1)) to yield **10** as a yellowish oil (4.26 g, 88%): IR (film) 3062, 2977, 2935, 1721, 1598, 1505, 1417, 1216, 1150, 1045, 978, 779, 752 cm⁻¹; ¹H NMR (250 MHz, CDCl₃) δ 1.23 (t, J = 7.1 Hz, 3H), 1.30 (t, J = 7.1 Hz, 3H), 3.43 (q, J = 7.1 Hz, 2H), 3.53 (q, J = 7.1 Hz, 2H, 7.24 (d, J = 8.7 Hz, 1H), 7.51 (td, J = 7.5, 1.1)Hz, 1H), 7.69 (td, J = 7.7, 1.5 Hz, 1H), 7.81 (d, J = 8.1 Hz, 1H), 8.01 (d, J = 8.4 Hz, 1H), 8.19 (d, J = 8.7 Hz, 1H); MS (EI, 70 eV) m/z 244 (18, M⁺), 227 (17), 145 (27), 128 (30), 116 (36), 100 (100), 72 (80).

N-(*N*,*N*-Diethylaminocarbonyl)-2-quinolone (11). Short path distillation (173–175 °C/0.5 Torr) of 4.26 g of 10 yielded a mixture of 10 and a new product. Flash chromatography on SiO₂ (Et₂O) gave recovered 10 (2.39 g) and a colorless crystalline material (1.81 g), which upon recrystallization from CHCl₃/Et₂O (1:3) yielded pure 11 (1.07 g, 25%): mp 108.5–110 °C; IR (KBr) 2988, 2943, 2880, 1695, 1659, 1591, 1430, 829 cm⁻¹; ¹H NMR (200 MHz, CDCl₃) δ 1.06 (t, J = 7.2 Hz, 3H), 1.41 (t, J = 7.2 Hz, 3H), 3.04–3.31 (m, 2H), 3.52–3.86 (m, 2H), 6.64 (d, J = 9.6 Hz, 1H), 7.13 (d, J = 8.5 Hz, 1H), 7.25 (td, J = 7.6, 1.0 Hz, 1H), 7.47–7.60 (m, 2H), 7.74 (d, J = 9.6 Hz, 1H); ¹³C NMR (50.3 MHz, CDCl₃) δ 12.6, 13.7, 41.8, 43.4, 114.4, 119.9, 121.6, 123.1, 128.6, 131.0, 137.3, 140.7, 152.0, 160.2; MS (EI,

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70 eV) m/z 244 (18, M⁺), 227 (9), 145 (12), 100 (100), 72 (67). Anal. Calcd for $C_{14}H_{16}N_2O_2$: C, 68.83; H, 6.60; N, 11.47. Found: C, 68.75; H, 6.70; N, 11.48.

N.N-Diethvl 1,2-Dihydro-2-oxo-3-quinolinecarboxamide (12). To a stirred LDA solution (diisopropylamine (592 mg, 5.9 mmol) and n-BuLi (3.9 mL (1.6 M in hexane), 6.2 mmol)) in THF (30 mL), a solution of 10 (1.10 g, 4.5 mmol) in THF (8 mL) (precooled to -78 °C) was added via cannula at -78 °C under N₂ atmosphere. After the solution was stirred for 1 h, anhydrous MeOH (562 mg, 17.5 mmol) was added and the reaction mixture was stirred for an additional hour at -78°C. After quenching with saturated aq NH₄Cl (15 mL), the aqueous phase was extracted with CHCl₃ (3 × 30 mL). The combined organic phase was washed with H₂O and dried (Na₂-SO₄). Removal of the solvent gave the crude product as yellow crystals (1.11 g). Flash chromatography on SiO₂ (EtOAc/MeOH (95:5)) yielded 12 as pale yellow crystals (666 mg, 61%): mp 166-167 °C (toluene) (lit. 31 mp 167 °C); IR (KBr) 3063, 2968, 2897, 1662, 1617, 1569, 1433, 1221, 946, 749 cm⁻¹; ¹H NMR (200 MHz, CDCl₃) δ 1.17 (t, J = 6.6 Hz, 3H), 1.32 (t, J = 6.6Hz, 3H), 3.34 (q, J = 6.8 Hz, 2H), 3.63 (q, J = 6.7 Hz, 2H), 7.24 (td, J = 7.5, 1.3 Hz, 1H), 7.41 (d, J = 7.9 Hz, 1H), 7.49 -7.60 (m, 2H), 7.87 (s, 1H), 12.36 (s br, 1H, exchangeable with

A second fraction was obtained and recrystallized from EtOAc/MeOH (2:1) to yield ${\bf 13}$ as colorless crystals (122 mg, 13%): mp >180 °C dec; IR (KBr) 3063, 2974, 2938, 2892, 2853, 1718, 1658, 1619, 1591, 1564, 1399, 1275, 1196, 1080, 754 cm $^{-1}$; 1 H NMR (200 MHz, CDCl $_{3}$ /DMSO- d_{6} (2:1)) δ 0.87 (t, J = 7.0 Hz, 3H), 0.94 (t, J = 7.0 Hz, 3H), 2.98 – 3.06 (m, 4H), 7.22 (t, J = 7.7 Hz, 1H), 7.39 (d, J = 8.2 Hz, 1H), 7.54 – 7.67 (m, 2H), 7.77 – 7.87 (m, 2H), 7.96 (d, J = 8.2 Hz, 1H), 8.12 (d, J = 7.9 Hz, 1H), 8.35 (s, 1H), 8.70 (s, 1H), 12.10 (s, 1H); 13 C NMR (50 MHz, CDCl $_{3}$ /DMSO- d_{6} (2:1)) δ 11.1, 11.8, 39.3, 39.7, 113.7, 116.7, 120.7, 124.8, 125.1, 125.9, 126.7, 127.2, 127.8, 129.0, 130.0, 130.9, 138.7, 138.8, 141.2, 145.0, 150.7, 151.8, 158.5, 188.7.

N,N-Diethyl 2-Trifluoromethanesulfonyloxy-3-quino**linecarboxamide (7).** To a solution of **12** (441 mg, 2.0 mmol) in CH₂Cl₂ (18 mL) at 0 °C was added NEt₃ (365 mg, 3.6 mmol) and Tf₂O (560 mg, 2.0 mmol) dropwise under N₂ atmosphere. The solution was stirred at 0 °C for 1 h and quenched with saturated aq NaHCO₃ (20 mL). The aqueous phase was extracted with CH_2Cl_2 (2 × 10 mL) and the combined organic phase was washed H₂O and dried (Na₂SO₄). Removal of the solvent gave the crude product as yellow crystals (887 mg). Flash chromatography on SiO₂ (Et₂O) afforded **7** as pale yellow crystals (478 mg, 70%): mp 89.5-91 °C (CHCl₃/hexane (4:1)); IR (film) 3059, 2981, 2939, 1635, 1573, 1424, 1217, 1140, 1052, 918, 884, 847, 801, 759 cm⁻¹; ¹H NMR (200 MHz, CDCl₃) δ 1.12 (t, J = 7.1 Hz, 3H), 1.31 (t, J = 7.1 Hz, 3H), 3.25 (q, J = 7.1 Hz, 3Hz)7.1 Hz, 2H), 3.61 (s br, 2H), 7.67 (td, J = 7.5, 1.2 Hz, 1H), $7.84 \text{ (td, } J = 8.4, 1.5 \text{ Hz, 1H)}, 7.90 \text{ (d, } J = 8.4 \text{ Hz, 1H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{ (d, } J = 8.4 \text{ Hz, 2H)}, 8.06 \text{$ J=8.5 Hz, 1H), 8.28 (s, 1H); $^{13}{\rm C}$ NMR (62.9 MHz, CDCl₃) δ 12.4, 13.9, 39.5, 43.1, 118.5 (q, 321), 122.6, 127.1, 127.8, 128.3, 128.6, 131.7, 139.0, 145.4, 149.6, 163.8; MS (EI, 70 eV) m/z 376 (36, M+), 304 (100), 227 (49), 172 (82), 143 (66); HRMS calcd for $C_{15}H_{15}F_3N_2O_4S$ 376.0705, found 376.0693.

N,N-Diethyl 2-(2-Pyridyl)-3-quinolinecarboxamide (14). A solution of 2-bromopyridine (117 mg, 741 μmol) in THF (3 mL) was treated at -78 °C with *t*-BuLi (1.7 M in pentane, 867 μL, 1.5 mmol) under N₂ atmosphere. After the dark red solution was stirred for 15 min at -78 °C, a solution of anhydrous ZnBr₂ (183 mg, 813 μmol) in THF (2 mL) was added via cannula. The resulting orange solution was stirred at -78 °C for 1 h and then allowed to warm to room temperature. A solution of 7 (185 mg, 492 μmol) and Pd(PPh₃)₄ (28 mg, 24 μmol) in THF (3 mL), which had been stirred for 15 min at room temperature, was added via cannula. The reaction mixture was refluxed (60 h) and quenched with saturated aq NH₄Cl (2 mL). The aqueous phase was extracted with CH₂Cl₂ (1 × 20 mL, 2 × 10 mL) and the combined organic phase was

washed with H₂O and dried (Na₂SO₄). Evaporation to dryness in vacuo gave the crude product as a brownish oil (208 mg), which was purified by flash chromatography on SiO₂ (CHCl₃) to yield 14 as a yellowish oil (85 mg, 57%): IR (CDCl₃) 3064, 2981, 2837, 2875, 1723, 1618, 1559, 1480, 1279, 1097 cm⁻¹; 1 H NMR (200 MHz, CDCl₃) δ 1.00 (t, J=7.1 Hz, 3H), 1.30 (t, J=7.1 Hz, 3H), 3.21 (s br, 2H), 3.93 (s br, 2H), 7.28–7.34 (m, 1H), 7.58 (t, J=7.2 Hz, 1H), 7.72–7.87 (m, 3H), 8.14 (s, 1H), 8.19 (d, J=8.5 Hz, 1H), 8.42 (d, J=7.9 Hz, 1H), 8.59 (s br, 1H); 13 C NMR (50 MHz, CDCl₃) δ 11.7, 13.3, 38.7, 42.7, 123.1, 123.8, 127.0, 127.4, 127.5, 129.6, 130.1, 130.7, 134.8, 136.6, 147.3, 148.2, 153.4, 156.2, 170.3; MS (EI, 70 eV) m/z (305 (<1, M+), 233 (100), 205 (19); HRMS calcd for $C_{19}H_{19}N_3O$ 305.1528, found 305.1500.

2-Bromo-6-methoxypyridine. A solution of NaOMe in MeOH (Na (1.33 g, 58 mmol) in anhydrous MeOH (14 mL)) was added to a suspension of 2,6-dibromopyridine (8.0 g, 34 mmol) in anhydrous MeOH (22 mL) and the resulting mixture was refluxed for 25 h. The reaction mixture was allowed to cool to room temperature, cold 5% aq NaHCO₃ (25 mL) was added, and the mixture was extracted with Et₂O (1 \times 30 mL, $2\,\times\,20$ mL). The combined organic phase was concentrated, the resulting residue was taken up in Et₂O (30 mL), and the Et₂O solution was washed with brine (20 mL) and dried (K₂-CO₃). Removal of the solvent gave a yellowish liquid (5.72 g). Kugelrohr distillation yielded 2-bromo-6-methoxypyridine as a colorless liquid (5.48 g, 86%): bp 87–91 °C/15 Torr (lit. 35 bp 85-95°/15 Torr); IR (film) 2985, 2952, 2850, 1594, 1581, 1556, $1467,\,1408,\,1296,\,1020,\,854\;\mathrm{cm^{-1}};\,{}^{1}\mathrm{H\;NMR\;(200\;MHz,\,CDCl_{3})}$ δ 3.93 (s, 3H), 6.68 (d, J = 7.7 Hz, 1H), 7.05 (d, J = 7.4 Hz, 1H), 7.40 (t, J = 7.8 Hz, 1H).

N,N-Diethyl 2-(2-Methoxypyridin-6-yl)-3-quinolinecar**boxamide** (6). To a solution of 2-bromo-6-methoxypyridine (301 mg, 1.6 mmol) in THF (6 mL) at -78 °C was added dropwise t-BuLi (1.7 M in pentane, 1.9 mL, 3.2 mmol) under N₂ atmosphere. The resulting pale yellow solution was stirred for 15 min at −78 °C. A solution of anhydrous ZnBr₂ (409 mg, 1.8 mmol) in THF (4 mL) was added via cannula and the mixture was stirred for 70 min at -78 °C. The reaction mixture was allowed to warm to room temperature and was added via cannula to a solution of 7 (399 mg, 1.1 mmol) and $Pd(PPh_3)_4$ (62 mg, 54 μ mol) in THF (4 mL), which had been stirred for 20 min at room temperature. The resulting solution was refluxed for 45 h and the solvent was removed in vacuo. The residue was dissolved in CHCl₃/MeOH (9:1) (15 mL) and washed with Na₂EDTA·2H₂O (1.35 g in 15 mL of H₂O). The aqueous phase was extracted with CHCl₃/MeOH (9:1) (5 \times 12 mL) and the combined organic phase was dried (Na₂SO₄). Removal of the solvent gave a yellow gum (569 mg) that was purified by flash chromatography on SiO₂ (EtOAc/hexane (2: 1)) to give 6 as a pale yellow oil (211 mg, 59%): IR (film) 3060, 2979, 2934, 2874, 1634, 1574, 1556, 1267, 1047, 921, 888, 792 cm $^{-1};$ $^{1}\mathrm{H}$ NMR (200 MHz, CDCl3) δ 0.84 (t, J=7.1 Hz, 3H), 1.22 (t, J = 7.1 Hz, 3H), 2.90 - 2.99 (m, 1H), 3.16 - 3.36 (m, 2H),3.79-3.87 (m, 1H), 3.97 (s, 3H), 6.79 (d, J = 8.2 Hz, 1H), 7.56(t, J = 7.9 Hz, 1H), 7.69-7.77 (m, 2H), 7.84 (d, J = 8.1 Hz, 1H), 7.97 (d, J = 7.4 Hz, 1H), 8.17 (s, 1H), 8.18 (d, J = 5.8 Hz, 1H); $^{13}{\rm C}$ NMR (50 MHz, CDCl $_{3}$) δ 12.8, 13.4, 38.9, 43.2, 53.7, $111.3,\ 116.3,\ 126.9,\ 127.4,\ 127.5,\ 129.5,\ 130.1,\ 130.3,\ 135.9,$ 139.2, 147.3, 153.4, 154.0, 163.4, 169.8; MS (EI, 70 eV) m/z $335 (7, M^+), 263 (100), 235 (23);$ HRMS calcd for $C_{20}H_{21}N_3O_2$ 335.1634, found 335.1602.

N,N-Diethyl 2-(2-Methoxypyridin-6-yl)-3-quinoline-thiocarboxamide (15). A solution of 6 (120 mg, 358 μmol) and Lawesson's reagent (241 mg, 596 μmol) in anhydrous benzene (3 mL) was refluxed for 8 h under N_2 atmosphere. The reaction mixture was passed through a cotton plug and the filtrate was concentrated. The resulting orange oil was purified by flash chromatography on SiO₂ (CH₂Cl₂/Et₂O (9:1)) to yield 15 as a yellow viscous liquid (105 mg, 83%): IR (film) 3064, 2982, 2941, 2876, 2221, 1592, 1575, 1498, 1266, 816 cm⁻¹; ¹H NMR (250 MHz, CDCl₃) δ 1.03 (t, J = 7.2 Hz, 3H),

1.27 (t, J = 7.1 Hz, 3H), 3.25 (dq, J = 14.3, 7.1 Hz, 1H), 3.66(dq, J = 14.3, 7.2 Hz, 1H), 3.95(s, 3H), 4.03-4.19(m, 2H),6.78 (dd, J = 8.0, 0.6 Hz, 1H), 7.56 (td, J = 8.0, 1.0 Hz, 1H),7.68-7.76 (m, 2H), 7.82 (dd, J=8.1, 0.7 Hz, 2H), 8.09 (s, 1H),8.17 (d, J = 8.5 Hz, 1H); ¹³C NMR (50 MHz, CDCl₃) δ 11.1, 13.3, 45.9, 48.7, 53.9, 110.9, 117.3, 126.8, 127.4, 127.5, 129.2, 130.1, 134.3, 135.8, 139.1, 146.7, 152.2, 154.2, 163.2, 197.6; MS (EI, 70 eV) m/z 351 (25, M⁺), 280 (47), 265 (100), 243 (12).

3-N,N-Diethylaminomethyl-2-(2-methoxypyridin-6-yl)**quinoline** (16). To a solution of 15 (105 mg, 299 μ mol) in anhydrous CH₂Cl₂ (1.5 mL) at 0 °C was added Et₃O⁺BF₄⁻ (0.5 M in CH_2Cl_2 , 0.62 mL, 310 μ mol) under N_2 atmosphere. The resulting mixture was stirred for 5 min at 0 °C and then for 90 min at room temperature. The solvent was removed in vacuo and the solid residue was dissolved in anhydrous MeOH (2.5 mL). NaBH₄ (29 mg, 767 μ mol) was added at 0 °C and the mixture was stirred for 5 min at 0 °C and then for 3 h at room temperature. HCl (5%, 2 mL) was added and the mixture was stirred for 5 min. After addition of aq NaOH solution to pH > 10, the reaction mixture was extracted with Et₂O (1 \times 20 mL, 2×15 mL). The combined organic phase was washed with H₂O (20 mL) and dried (Na₂SO₄). Removal of the solvent gave a yellow oil (101 mg) that was purified by flash chromatography on SiO₂ (EtOAc) to yield **16** as a yellow gum (80 mg, 83%): IR (CDCl₃) 3064, 2971, 2935, 2874, 2809, 1722, 1576, 1464, 1411, 1261 cm $^{-1}$; ¹H NMR (250 MHz, CDCl₃) δ 0.98 (t, $J = 7.1 \; \mathrm{Hz}, \, 6\mathrm{H}), \, 2.54 \; (\mathrm{q}, \, J = 7.1 \; \mathrm{Hz}, \, 4\mathrm{H}), \, 3.96 \; (\mathrm{s}, \, 3\mathrm{H}), \, 4.09 \; (\mathrm{s}, \, 3\mathrm{H$ 2H), 6.80 (dd, J = 8.4, 0.6 Hz, 1H), 7.49 - 7.55 (m, 2H), 7.64 -7.77 (m, 2H), 7.86 (d, J = 8.1 Hz, 1H), 8.14 (d, J = 8.5 Hz, 1H), 8.53 (s, 1H); 13 C NMR (63 MHz, CDCl₃) δ 11.8, 47.2, 53.4, 54.9, 110.2, 117.3, 126.6, 127.4, 127.9, 129.0, 129.2, 132.3, 136.7, 139.3, 156.8, 157.2, 162.7; MS (EI, 70 eV) m/z 321 (35, M⁺), 292 (100), 249 (86), 235 (50), 205 (40), 185 (53); HRMS calcd for $C_{20}H_{23}N_3O$ 321.1841, found 321.1820.

11*H*-Indolizino[1,2-*b*]quinolin-9-one (5). To a suspension of anhydrous K₂CO₃ (23 mg, 166 μmol) and ClCOOEt (25 μL,

261 µmol) in anhydrous THF (1 mL) was added a solution of 16 (67 mg, 208 μ mol) in anhydrous THF (1 mL) via cannula under N₂ atmosphere. The mixture was stirred at room temperature for 2.5 h. TLC showed a new spot with intense blue fluorescence under UV light. The reaction mixture was heated at reflux for 3 h. To the resulting orange suspension was added ClCOOEt (10 μ L, 105 μ mol) and the mixture was heated at reflux for 1 h. After the solution was cooled to room temperature, H₂O (10 mL) was added and the whole was extracted with CH₂Cl₂ (3 × 10 mL). The combined organic phase was washed with brine (10 mL) and dried (Na₂SO₄). Removal of the solvent gave a yellow solid (48 mg) that was purified by flash chromatography on SiO₂ (EtOAc/NEt₃ (99: 1)) to yield $\bf 5$ as pale yellow crystals (30 mg, 62%) (lit. 27a mp 265 °C; lit. 27c mp 253-254 °C): IR (CDCl₃) 3056, 1663, 1594, 1159 cm $^{-1};$ ^1H NMR (250 MHz, CDCl_3) δ 5.25 (s, 2H), 6.73 (dd, J = 9.0, 0.7 Hz, 1H, 7.30 (dd, J = 6.1, 0.8 Hz, 1H), 7.60-7.70(m, 2H), 7.80 (ddd, J = 8.4, 7.0, 1.4 Hz, 1H), 7.89 (d, J = 8.1Hz, 1H), 8.20 (d, J = 8.5 Hz, 1H), 8.34 (s, 1H). The spectroscopic properties are identical with those reported for 5 prepared previously.²⁷

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Supporting Information Available: Spectroscopic information and reagent purification and availability. This material is available free of charge via the Internet at http://pubs.acs.org.

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